

imee[®]

BATTERY

R6P 1.5V AA CARBON ZINC BATTERY
R03 1.5V AAA CARBON ZINC BATTERY



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#AskForimee

DECADES OF TRUST, INNOVATION, SUPPORT & EXCELLENCE

Introduction...

iMee LED is a brain child of Parekh Group, established in 1945 with a vision of honesty, truthfulness and quality. The Parekh Group is counted among the leading business houses having diversified businesses across various sectors with an expertised experience of over seven decades and four generations.

Owing to our rich expertise, knowledge, experience and expertise of our expert team, "we offer a wide range of LED Lights, LED TV's, Personal & Home care products and much more range". Our entire product range is under strict supervision of our team with utmost attention to intricate detail so that the best products and services can be conveyed to our clients.

Our team consists of highly skilled and professionally trained people who work hard in close coordination with each other to set up total customer satisfaction. Through an unwavering focus on commitment & delivery, we have charted our way to success in our operations and have won the admiration of our buyers and have built a long term relationship with our esteemed customers.

Assurance of Quality & Excellence has been our key to success since our establishment and we believe that the years ahead will see us treading new grounds, widening the spectrum and expanding our trade horizons even further. We have set a high standard of conduct towards our buyers. Our precise approach to the quality of our manufactured products meets our customers' demands. We continuously work on improvement of our products, services & ourselves, so that we are able to serve you in the best possible way; today and for years to come.

Brand Owned By:

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BATTERIES



R6P 1.5V AA CARBON ZINC BATTERY

- R6P 99A+GT Zinc Carbon Battery
- Guaranteed Leak Proof Up to 0.6 CCV
- No Mercury Added
- Heavy Duty & Leak Proof
- Aluminum Jacket
- Voltage : 1.5V
- Temperature Range: -25° C to 55° C

R03 1.5V AAA CARBON ZINC BATTERY

- R03 99A+GT Zinc Carbon Battery
- Guaranteed Leak Proof Up to 0.6 CCV
- No Mercury Added
- Heavy Duty & Leak Proof
- Aluminum Jacket
- Voltage : 1.5V
- Temperature Range: -25° C to 55° C



imee Carbon Zinc (Zn/MnO_2) Application Manual

imee Carbon Zinc batteries are marketed in three basic grades— Super Heavy Duty, Heavy Duty and General Purpose. Super Heavy Duty is the premium Carbon Zinc, which performs better on moderate to heavy drains or continuous drains versus the Heavy Duty and General Purpose grades.

Carbon Zinc batteries come in two electrochemical systems: the LeClanche and Zinc Chloride. Most cylindrical **imee** Carbon Zinc batteries are constructed using the Zinc Chloride system with two different sealing and jacket constructions namely: Paper Jacket and Metal Jacket.

All **imee** make carbon zinc batteries have no added mercury or cadmium.

System Description:

Carbon Zinc: A generic term for primary dry batteries of Zinc Chloride system. This battery has an anode of zinc, a cathode of manganese dioxide, and a slightly acidic electrolyte.

Carbon Zinc batteries provide an economical power source for devices requiring light to moderate drain because of the use of inexpensive materials and their time proven constructions. All **imee** battery of carbon zinc cells are primary batteries and therefore are not designed for recharging.

The service capacity of a Carbon Zinc battery is not a fixed number of ampere hours because the battery functions at different efficiencies depending upon the conditions imposed upon it. The service varies with current drain, operating schedule, and cutoff voltage. The battery is also affected by the operating temperature and storage conditions.

The general characteristics of a **imee** make Zinc Chloride battery are:

- Less expensive than alkaline. Economical in terms of cost per hour on moderate current drains or use frequency.
- Less output capacity decrease than LeClanche as the drain rate increases.
- Less sensitive than LeClanche to changes in the discharge rate or duty cycle.
- Lower internal resistance than LeClanche.
- Better low temperature performance than LeClanche.
- Energy density of approximately 2 to 2.5 watt hours per cubic inch.
- Average service maintenance exceeds 90% after one year storage at 21°C on typical tests.
- Higher open circuit and initial closed circuit voltage than LeClanche or alkaline.
- Lower unit weight than alkaline.
- Available in voltages ranges of 1.5 volts and in a variety of shapes and sizes.
- Sloping discharge curve.



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Carbon Zinc Battery

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The carbon zinc battery uses a zinc anode, a manganese dioxide cathode, and an electrolyte of zinc chloride dissolved in water. Powdered carbon is used in the cathode mix, usually in the form of carbon black to improve conductivity of the mix and for moisture retention.

Carbon zinc batteries are produced in general configurations:

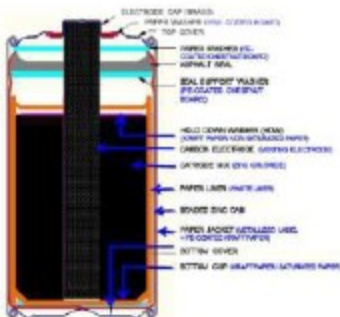
- Cylindrical--available as unit cells

Within the carbon zinc cylindrical battery category are two constructions: Paper Jacket and Metal Jacket. The Zinc Chloride battery contains proportionately zinc chloride in the electrolyte and therefore requires different battery design as shown in the following diagram:



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PAPER JACKET CONSTRUCTION



METAL JACKET CONSTRUCTION



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Carbon Zinc Battery

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Cathodes are a mixture of manganese dioxide, carbon conductor and electrolyte.

Anodes are zinc alloy can. The can also confines the active materials in the battery.

Separators are coated paper selected to prevent migration of solid particles in the battery.

Carbon electrode serves as the cathode current collector.

Top and bottom covers provide contact surfaces of plated steel.

The outside of the battery is either covered with plastic film and metal jacket or Kraft paper and a printed paper label.

Asphalt or plastic seal acts as the battery seal.

Electro-Chemistry:

LeClanche

The performance of a LeClanche battery is the result of an electrochemical reaction between:

A cathode composed of carbon and refined manganese dioxide which may contain some naturally occurring manganese dioxide. The more pure the cathode material, the better the performance. (The carbon component of the cathode is usually carbon black and provides increased conductivity and moisture retention.)

An anode of high purity zinc alloy.

A highly conductive, slightly acidic, electrolyte solution of ammonium chloride and zinc chloride in water.

The chemical equation for this reaction is:



Zinc Chloride

The performance of a Zinc Chloride battery is the result of an electrochemical reaction between:

A cathode composed of carbon and refined manganese dioxide which may contain some naturally occurring manganese dioxide.

- The carbon component of the cathode is usually carbon black and provides increased conductivity and moisture retention.
- Typically, Zinc Chloride batteries have a higher proportion of carbon to manganese dioxide than LeClanche.

An anode of high purity zinc alloy.

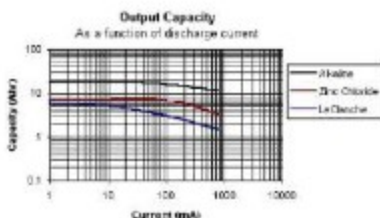
A highly conductive, slightly acidic, electrolyte solution of zinc chloride in water which may contain a small amount of ammonium chloride.

- A Zinc Chloride battery contains a greater volume of electrolyte than the same size LeClanche battery.
- The electrolyte is slightly more acidic than a LeClanche electrolyte.

The chemical equation of this reaction is:



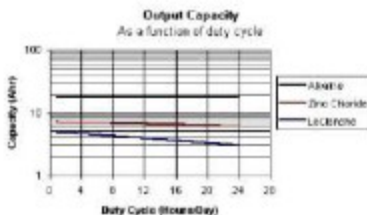
A Zinc Chloride battery is typically over 1.60 volts. The closed circuit voltage declines gradually as a function of the depth of discharge. The energy output of Zinc Chloride batteries is less sensitive to variations in the discharge current and duty cycle than comparable size LeClanche batteries. Typical D size performance to a 0.75 volt cutoff is shown in the following diagrams:



The efficiency of a carbon zinc or alkaline battery improves as the current drain decreases as seen in the above graph. As a result, an important application guide-line should be considered: "For increased efficiency, use as large a battery as possible, consistent with the physical limitations of the device." This has the same effect as lowering the current. As an example, doubling the size of a carbon zinc battery will more than double the service life at a given drain.



ALKALINE BATTERY



ZINC CHLORIDE BATTERY

The electrochemical inputs of cylindrical D size batteries typically are in a ratio of 2:3:5 for, Zinc Chloride and Alkaline respectively. The differences in efficiency and rate sensitivity between the two systems cause variations in actual output in simulated typical applications as shown in the following table and graph:

Carbon Zinc Battery

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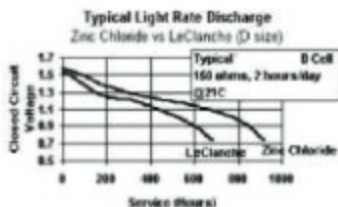


D Size

Typical Percent of Zinc Chloride vs Alkaline Service

Test	Load (Ohms)	Duty Cycle	Zinc Chloride	Alkaline
Motor Toy	2.2	1Hpd to 0.8 V	100%	250%
Flashlight	2.2	4 min/hr 8 hr/day to 0.9V	100%	300%
Radio	24	4 hr/day to 0.9 V	100%	260%

Carbon Zinc batteries are more efficient when used in low rate applications as shown in the curve below. Typical carbon zinc light drain is defined as a current that would discharge the battery after 50 or more hours of use at room temperature.



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As the drain is increased, the service difference between Alkaline and Zinc Chloride systems increases. This relationship is shown by the following discharge curves. Typical carbon zinc moderate drain is defined as a current that would discharge the cell within 10-50 hours of use at room temperature.

Typical Moderate Rate Discharge Zinc Chloride vs. LeClanche (D size)



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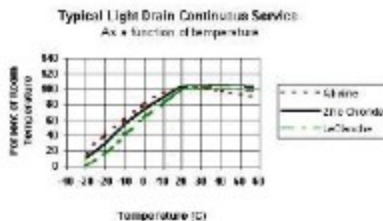
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Temperature

Changes in temperature will affect the reactivity of battery chemical components. The typical effect on service of a D size cylindrical battery to a 0.75 volt cutoff is shown in the following diagram:



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Light Drain is defined as a current that would discharge the battery after 50 or more hours of use at room temperature.

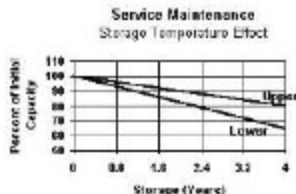
Heavier Drains at low temperature will tend to decrease the percent service from that shown in the above diagram. Zinc Chloride system is affected moderately, and Alkaline the least as the drain increases. The service on all drains at high temperatures over time is eventually reduced by an increase in self discharge.

Carbon Zinc batteries provide good service maintenance due to time tested construction, quality control of materials and close monitoring of batteries during assembly.

Service Maintenance (storage at 21°C)

Time of Storage at 21°C	Typical % of Fresh Service Retained
1 year	90 - 100%
2 years	80-90%
3 years	75-85%
4 years	65-80%

The storage of carbon zinc batteries at temperatures below 21°C will increase their service maintenance. While freezer storage (-20°C) of a carbon zinc battery is not harmful, storage at 5 to 10°C is effective. Batteries to be stored at low temperature storage should be allowed to reach room temperature in their packing so as to avoid condensations of moisture which may cause electrical leakage and/or destruction of the jackets. Storage at high temperatures exceeding 21°C for sustained periods of time will significantly reduce service maintenance. The typical effects of storage temperature on carbon zinc service maintenance are shown in the following diagram:



Internal Resistance:

The internal resistance (R_i) of a battery is its opposition to the flow of current. In all cases, this resistance increases as the temperature of a battery decreases. While the R_i will vary with load for the battery size, it will be higher for LeClanche than Zinc Chloride which in turn will also be higher than Alkaline. The R_i of a cylindrical carbon zinc battery increases gradually until it approaches the end of service life and then increases rapidly.

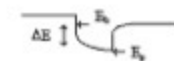
Internal resistance is typically measured in one of two ways:

1. As a reduction in closed circuit voltage when the applied load is increased (voltage drop method).
2. As a maximum short circuit current (flash amperage).

The voltage drop method in determining the effective internal resistance is also used by ANSI.

The R_i values obtained by either method of measurement are subject to number of variables and operator techniques. The effective R_i values shown on the data pages were calculated by the voltage drop method as this more accurately projects the batteries current carrying capability in actual device applications. This calculation involves placing a battery on a constant background load, allowing it to stabilize and then pulsing it with a heavier load for one second. The resulting voltage drop is then measured and expressed in terms of Ohms as shown in the following example.

Determination of Internal Resistance (R_i)



$$I_1 = \frac{E_b}{R_b}$$

$$I_2 = \frac{E_b}{R_b}$$

$$R_i = \frac{\Delta E}{\Delta I} = \frac{E_1 - E_2}{I_1 - I_2}$$

Voltage Drop Method

R_i = Internal Resistance

R_b = Resistance of Background Load



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- E_b = Background Voltage
- R_p = Resistance at Pulse Load
- E_s = Voltage at end of pulse
- E = Voltage Change
- I = Current Change
- i_b = Background Current
- i_p = Current at End of Pulse

Although flash amperage does not indicate battery freshness or potential service, circuit designers should be aware of the maximum current that a battery could supply if a component failure occurs. The following are typical maximum flash amperage values for **imee** carbon zinc batteries. These flash amperage values can vary widely without affecting battery service in actual applications.

Typical Maximum Flash Amperage for CZ Zinc Chloride Battery	
D	9
AA	5
AAA	3

Applications

imee carbon zinc batteries will meet a wide variety of device applications utilizing light to moderate drains, such as:

- Alarm Systems
- Calculators
- Clocks
- Communications equipment
- Electronic games
- Flashlights
- Garage door openers
- Home entertainment remote controls
- Home security devices



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- Laboratory instruments
- Lanterns
- Motor driven devices
- Penlights
- Personal care devices
- Portable tape recorders and players
- Radios
- Radio controlled toys
- Remote control transmitters
- Small lighted toys and novelties
- Stereo headsets
- Test equipment
- Toys



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